

Kalorische Zustandsgl.

$$u = u(T, v)$$

$$du = \left(\frac{\partial u}{\partial T}\right)_v dT + \left(\frac{\partial u}{\partial v}\right)_T dv$$

$$= c_v dT + \left(\frac{\partial u}{\partial v}\right)_T dv$$

$$h = h(T, p)$$

$$dh = c_p dT + \left(\frac{\partial h}{\partial p}\right)_T dp$$

Wärmezufuhr:

$$Q_{12} = m \bar{c}_v \Big|_{T_1}^{T_2} (T_2 - T_1)$$

isochor + reversibel

$$Q_{12} = m \bar{c}_p \Big|_{T_1}^{T_2} (T_2 - T_1)$$

isobar + reversibel

$$\bar{c} \Big|_{T_1}^{T_2} = \frac{1}{T_2 - T_1} \int_{T_1}^{T_2} c dT$$

Berechnung von  $\bar{c} \Big|_{T_1}^{T_2}$  aus Tabellenwerten  $\bar{c} \Big|_{T_0}^T$

# Aufgabe 7b

geschätzt:  $\bar{c}_p \Big|_{0^\circ\text{C}}^{220^\circ\text{C}} \approx 0,930 \frac{\text{kJ}}{\text{kg}\cdot\text{K}}$

T	$c_p$
100	0,908
220	
300	0,954

$$y = 0,908 + \frac{120}{200} (0,954 - 0,908)$$

$$\bar{c}_p \Big|_{0^\circ\text{C}}^{225^\circ\text{C}} = 0,9356 \frac{\text{kJ}}{\text{kg}\cdot\text{K}}$$