

Aufgabe 4

$$1 \text{ eV} = 1.602 \cdot 10^{-19} \text{ J}$$

a.) $E_{\text{kin}} = 3 \cdot 10^{20} \text{ eV} = 11.49 \text{ Cal} \Rightarrow$ erwärmt $11.49 \text{ g Wasser um } 1^\circ \text{ C}$

b.) $E = E_0 + E_{\text{kin}} = \gamma m_0 c^2 = \gamma E_0$

$$\Rightarrow \gamma = 1 + \frac{E_{\text{kin}}}{E_0} = 3.198 \cdot 10^{11}$$

$$(E_0 = 938 \text{ MeV} = 938 \cdot 10^6 \text{ eV})$$

$$\gamma = \frac{1}{\sqrt{1-\beta^2}} \Rightarrow \beta = \sqrt{1 - \frac{1}{\gamma^2}} = \left(1 + \left(-\frac{1}{\gamma^2}\right)\right)^{1/2} \approx 1 + \frac{1}{2} \left(-\frac{1}{\gamma^2}\right) = 1 - \frac{1}{2\gamma^2}$$

$$\Rightarrow 1 - \beta = \frac{1}{2\gamma^2} = 4.888 \cdot 10^{-24}$$

c.) $v \approx c \Rightarrow \Delta t_E = 25000 \text{ a}$

$$\Delta t_p = \frac{1}{\gamma} \Delta t_E = 2.467 \text{ s}$$

$$1 \text{ a} = 60 \cdot 60 \cdot 24 \cdot 365,2425 \text{ s}$$